

Hetero [6+3] Cycloaddition of Fulvenes with *N*-Alkylidene Glycine Esters: A Facile Synthesis of the Delavayne and Incarvillateine Framework

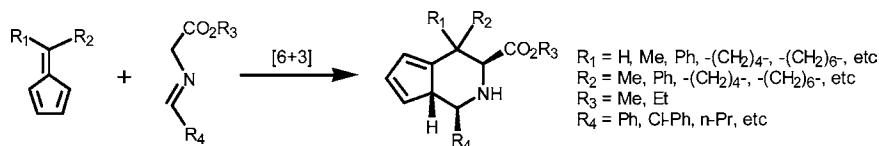
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ABSTRACT



In contrast to the [3+2] or [4+3] cycloaddition of *N*-metalated azomethine ylides and various alkenes, *N*-benzylidene glycine ethyl ester reacts with fulvenes to give the hetero [6+3] cycloaddition adducts with high stereoselectivity, constituting an efficient and novel route to [2]-pyridines.

The theoretical, mechanistic, and synthetic importance of fulvene and its derivatives have intrigued chemists for more than a century.¹ Cycloadditions of fulvenes (e.g. [4+3],² [2+2],³ [4+2],⁴ [2+4],⁵ [6+4],⁶ [6+2]⁷) provide versatile and powerful approaches to various polycyclic systems and natural products. Recently, we reported a new type of reaction: the [6+3] cycloaddition of fulvenes⁸ for the facile

synthesis of indan derivatives.⁹ More recently, Barluenga et al. demonstrated that the [6+3] cycloaddition of chromium alkenyl carbene complexes with fulvene leads to indanes.¹⁰ Additionally, we recently reported a novel hetero [6+3] cycloaddition of fulvenes for the synthesis of 11-oxa-steroids.¹¹ In conjunction with our continuing efforts in fulvene chemistry,¹² we have now developed a hetero [6+3] cycloaddition of fulvenes and *N*-benzylidene glycine ethyl ester that yields [2]pyridines. To the best of our knowledge,

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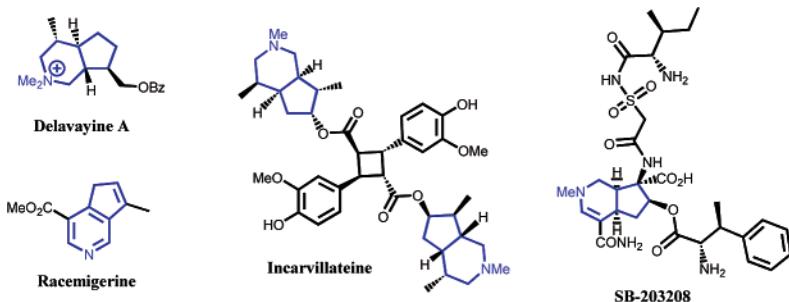
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Scheme 1



the synthesis of [2]pyridines via a hetero [6+3] cycloaddition has never been reported. [2]Pyridine systems can be found in a variety of natural products including delavayne A,¹³ SB-203208,¹⁴ incarvillateine,¹⁵ louisianin A,¹⁶ and racemigerine¹⁷ (Scheme 1).¹⁸ The 1,3-dipolar cycloaddition of *N*-alkyl glycine ester to alkenes via a [3+2] pathway¹⁹ or with a diene via a [4+3] pathway²⁰ represents an efficient and convergent approach to pharmacologically active alkaloids (e.g. the synthesis of pyrrolidines²¹ via the [3+2] cycloaddition reaction of azomethine ylides²² and alkenes). The 1,3-dipolar cycloaddition of fulvene has received much less attention, but examples of the [6+4], [4+2], and [3+2] cycloadditions of fulvene have been reported.^{23–25}

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On the basis of our previous observations, we suspected that the addition of a heterodipolar reagent, such as an azomethine ylide, to fulvene could afford the hetero [6+3] cycloadduct and provide a novel route to the [2]pyridine skeleton. In a model study, we have found that the *N*-benzylidene glycine ethyl ester derived from benzaldehyde and glycine ethyl ester in the presence of LDA in dry THF reacts with 6,6-dimethylfulvene (**1**) to yield the predicted hetero [6+3] cycloadduct **4** as the only isolable product in 80% yield (Scheme 2). The structure of **4** was assigned based on IR, ¹H, ¹³C NMR, COSY, DEPT, HMQC, HMBC, MS, and HRMS analysis. The formation of **4** may be rationalized via the stepwise mechanism shown in Scheme 2. Initial addition of the metalloazomethine ylide **2** to the C-6 position of fulvene **1** generates the zwitterionic intermediate **3**. This is followed by cyclization to give the [2]pyridine **4**. The chairlike transition state places the alkyl substituents at the equatorial positions throughout the cyclization process and leads to the formation of adduct **4** with high stereoselectivity. The azomethine ylides were generated by using a variety of methods (Table 1, entry 1, methods B–F). Among these, method D (Ag₂O in Et₃N–THF) gave the highest yield (92%) along with 8% of the uncyclized imine.

A series of homologous metalloazomethine ylides were then reacted with various fulvenes to afford the corresponding products **6**, **8**, **10**, and **12** (entries 2–5, Table 1).²⁶ The structure of **8** was unambiguously assigned by single-crystal X-ray analysis (Figure 1).²⁷ The reaction of various monoalkylfulvenes with metalloazomethine ylides gave similar adducts **14**, **17** and **15**, **18** in a 1:1 ratio of stereoisomers, respectively (entries 6–7, Table 1). The structure of **14** was also unambiguously assigned by single-crystal X-ray analysis (Figure 1).²⁸

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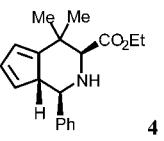
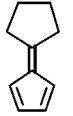
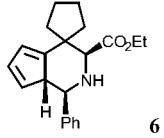
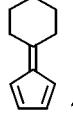
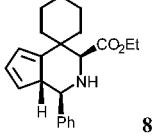
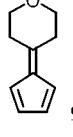
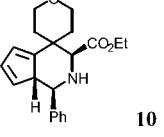
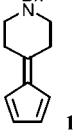
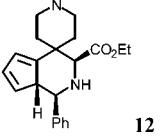
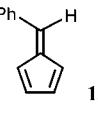
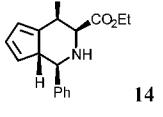
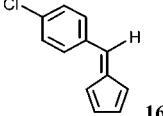
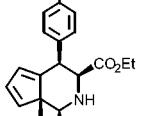
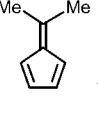
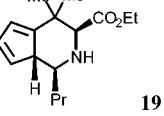
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(26) All new compounds were fully characterized by ¹H NMR, ¹³C NMR, DEPT, IR, MS, and HRMS. In most cases COSY and HMQC spectra were also obtained. Yields refer to spectroscopically and chromatographically homogeneous (>95%) materials.

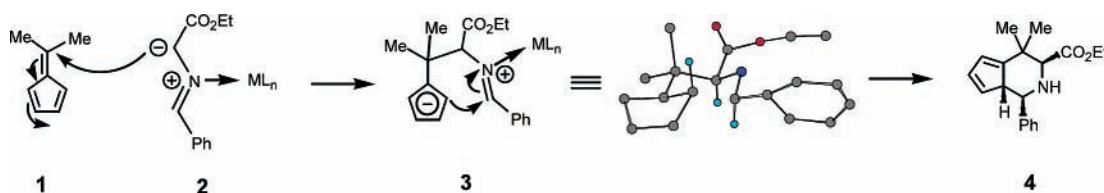
(27) Crystallographic data for **8**: C₂₂H₂₇NO₂, M = 337.45, monoclinic, space group *P*₂/*c*, *T* = 295 K, *a* = 8.2285(1) Å, *b* = 23.0207(4) Å, *c* = 10.2019(2) Å, β = 99.0300(6) $^\circ$, *V* = 1908.55(6) Å³, *Z* = 4, *D* = 1.174 g/cm³, λ (Mo $\text{K}\alpha$) = 0.71073 Å, 13582 reflections collected, 4381 unique reflections, 227 parameters refined on *F*², *R* = 0.0669, *wR*[*F*²] = 0.1773 [2341 data with *F*² > 2σ(*F*²)].

Table 1. Reaction of *N*-Alkylidene Glycine Ester with Fulvenes

entry	fulvene	product	method	time (h)	yield (%) ^a
1			A	1	80
			B	24	75
			C	24	20
			D	12	92 ^b
			E	12.5	7 ^c
			F	4	53 ^d
			G	6 for step 1 4 for step 2	75 ^b
2			A	1	57
			D	12	70 ^b
3			A	1	73
			D	12	86 ^b
4			A	1	66
			D	12	78 ^b
5			A	1	75
			D	12	89 ^b
6			A	1	71
			D	12	63 ^b
7			A	1	74
			D	12	68 ^b
8			G	6 for step 1 4 for step 2	67 ^{e,f}

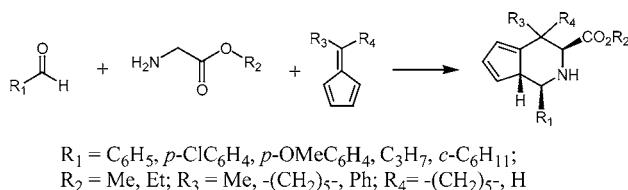
^a Isolated yield based on starting fulvene. Method A: LDA, THF, -78 °C. Method B: LiBr, Et₃N, THF, 25 °C. Method C: toluene, reflux. Method D: Ag₂O, Et₃N, THF, 25 °C. Method E: LiBr, DBU, 25 °C. Method F: AgOAc, Et₃N, 25 °C. Method G: glycine ethyl ester, C₆H₅CHO, MgSO₄, toluene, reflux, 12 h; fulvene 1, Ag₂O, Et₃N, 25 °C, 12 h. ^b 8% of the uncyclized imine was obtained. ^c 90% of the uncyclized imine was obtained. ^d 47% of the uncyclized imine was obtained. ^e Reacted with *N*-propyl glycine ethyl ester hydrochloride. ^f Total yield for two steps.

Scheme 2



The two-step reaction can be carried out in one pot by heating a 64 mM solution of benzaldehyde (1 equiv), glycine ethyl ester hydrochloride (1.3 equiv), Et_3N (5 equiv), and $MgSO_4$ in toluene to reflux for 6 h, followed by addition of

Scheme 3



a THF solution of fulvene **1** (1.2 equiv), Et_3N , and Ag_2O at ambient temperature and stirring for 4 h (Table 1, entry 1, method G, Table 1). This process yields adduct **4** in 75% yield without the need for isolation of the *N*-alkylidene glycine ester.

Next a selection of 3 fulvenes, 2 glycine esters, and 5 aldehydes were reacted according to Method G to yield a 30-membered [2]pyrindine library. During this process, heating in toluene was maintained for 12 h and the cyclization was allowed to proceed at ambient temperature for 8 h. Simple filtration through Celite and removal of the solvent afforded the final products in good yield and pure enough for MS and/or NMR analysis without further purification.

(28) Crystallographic data for **14**: $C_{23}H_{23}NO_2$, $M = 345.42$, monoclinic, space group $P2_1/c$, $T = 295$ K, $a = 11.0990(9)$ Å, $b = 8.6516(7)$ Å, $c = 20.1131(16)$ Å, $\beta = 101.3730(10)^\circ$, $V = 1893.43(3)$ Å 3 , $Z = 4$, $D = 1.212$ g/cm 3 , λ (Mo $K\alpha$) = 0.71073 Å, 8110 reflections collected, 2732 unique reflections, 237 parameters refined on F^2 , $R = 0.0473$, $wR2[F^2] = 0.1338$ [2339 data with $F^2 > 2\sigma(F^2)$].

In summary, we have developed a novel synthesis of [2]-pyridine derivatives (delavayine and incarvillateine skeletons) via a stereoselective one-pot hetero [6+3] cycloaddition of *N*-alkylidene glycine esters to fulvenes. We are currently pursuing the application of this methodology to the solid-phase synthesis of a large [2]pyridine library and other natural products.

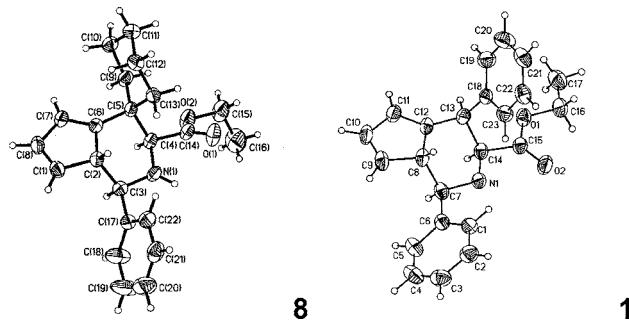


Figure 1. ORTEP plots for X-ray crystal structures of **8** and **14**.

Acknowledgment. We are grateful to Dr. Sepehr Sarshar for valuable discussions. Financial support from National Science Council and National Health Research Institute are gratefully acknowledged.

Supporting Information Available: Crystallographic information files (CIF) for **8** and **14** and experimental procedures and characterization data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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